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Dehydrophenylalanine (Δ Phe) as a β Breaker: Extended Structure Terminated by a Δ Phe-Induced Turn in the Pentapeptide Boc-Phe1-Ala2-Ile3-APhe4-Ala5-OMe

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Amyloid is a highly insoluble, aggregated state of certain polypeptide sequences associated with a range of debilitating diseases. A key step in amyloid formation is the transition of a protein from its native structure to a β -sheet arrangement; this suggests that the prevention of the ability of amyloidogenic proteins to adopt a β -sheet conformation would be useful as a way to impede the amyloid self-assembly process.^[1] The use of β -breaker residues is one approach for the development of peptide-based fibrillization-inhibiting drugs. Soto et al. demonstrated that the incorporation of β -sheet-breaker elements into short peptides composed of the recognition sequence of the amyloidogenic proteins inhibited amyloid formation.^[1c, d] In this context, β -sheet-breaker residues, such as proline and α -aminoisobutyric acid (Aib), which is an unnatural amino acid residue, have been found to inhibit amyloid fibril formation.^[2a-d]

 α , β -Dehydrophenylalanine (Δ Phe) is an analogue of phenylalanine, but it has a double bond between the $Ca = CB$ atoms, which leads to an extended conjugation with the π electrons of the ring; this makes Δ Phe a planar residue. Δ Phe induces β turns in short peptides (even as short as tripeptides) and 3_{10} helices in long peptides.^[3a–e] In this regard Δ Phe and Aib show similar conformational behavior.^[2c, 3d] The potential of Aib as a β -sheet-breaker residue is already known.^[2a-c] Thus, the present study is an endeavor to explore the potential of Δ Phe as a β -sheet-breaker residue. Earlier studies from our laboratory have shown that the incorporation of Δ Phe along with protein-coded helix-forming residues induces $3₁₀$ helical structures.^[3e] In this study, a peptide, Boc-Phe1-Ala2-Ile3- Δ Phe4-Ala5-OMe (1), which contains two β -strand-preferring amino acids (Phe1, Ile3), $[4]$ was synthesized, and its molecular conformation was studied by using X-ray crystallography (Figure 1). Table S1 in the Supporting Information shows the crystallographic details of the peptide. The stretch, Phe1-Ala2-Ile3, has

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Figure 1. The molecular conformation of peptide 1 (Boc-FAI Δ FA-OMe).

an extended (β strand) conformation, and has average ϕ, ψ values of $(-125^{\circ}, 114^{\circ})$. By assuming a helical conformation, however, Δ Phe4 terminated strand propagation. The important torsion angles for the peptide are shown in Table S2. The tertbutoxycarbonyl (Boc) group conformation is characterized by C1-O1-C5-N1 (θ^1) and O1-C5-N1-C1A (ω_o) torsion angles. It is generally noted that Boc groups with a trans–trans conformation permit the carbonyl oxygen O2 to act as an acceptor for the first $4 \rightarrow 1$ hydrogen bond in helical peptides.^[5] In the present peptide, Boc has a trans–trans conformation $(\theta^1 = -176.1$ $(3)^\circ$, $\omega_0 = -166.9$ (3)^o), and O2 is involved in a N-H···O intermolecular hydrogen bond with a symmetry-related parallel strand in the crystal lattice. All the peptide units are in the trans conformation (Table S2). As expected, the Δ Phe4 side chain is planar, as indicated by the angles $\chi^1=-5^\circ$ and $\chi^2=-24^\circ$ (Table S3). It is noteworthy that chain termination and turn formation occur together with the presence of C-H-··O hydrogen bond between atoms $C\gamma$ 2 of Ile3 and carbonyl oxygen of Ala5 (Table S3). In the crystal packing, β strands are arranged both in parallel and antiparallel orientations (Figure 2 A and B). The β strands extend approximately along the crystallographic c axis. Parallel β strands are related by translation symmetry along the crystallographic *a* axis and they define β sheets in the a–c plane, which are stabilized by a network of interstrand N-H \cdot ^{-O} and C α -H \cdot -O hydrogen bonds (Figure 3). The pattern of interstrand hydrogen bonds is similar to that seen between parallel β strands in proteins.^[6] Antiparallel strands are related to each other by a $2₁$ screw axis along the b direction, and they are stabilized mainly through aromatic–aromatic and van der Waals interactions. (The results of a search carried out in the PDB database for proteins that contain side-chain to mainchain hydrogen bonds, similar to the arrangement seen in the

Figure 2. A) Parallel and B) antiparallel orientation of β strands in the crystal packing of peptide 1.

Figure 3. The network of N-H- \cdot -O and C α -H \cdot -O hydrogen bonds between parallel β strands in the crystal packing of peptide 1.

present pentapeptide, is shown in Table S4.) The present peptide provides the first example of an extended structure that is terminated by a Δ Phe residue; this led us to design peptidebased inhibitors of amyloid-fibril formation.

Alzheimer's disease is the most common form of dementia and is associated with the deposition of senile plaques that are formed by the aggregation of β -amyloid polypeptide (39–

42).^[7a, b] Through peptide array technology, a central region $(AB₁₆₋₂₀)$ that mediates intermolecular interactions between $A\beta_{1-40}$ (A β) monomers to form amyloid fibrils, has been identified.^[1g] Short synthetic peptides that contain β -sheet-breaker residues have been shown to inhibit Aß-induced fibrillization.^[1c,f,7c] In this context, the specific role of Δ Phe as a β sheet-breaker residue was investigated by incorporating it in the 16–20 segment of the A β sequence. Three peptides, AcKLVFFNH₂ (2; A β_{16-20} with N- and C-terminal acetylation and amidation), AcKL Δ FVFNH₂ (3), and AcKLV Δ FFNH₂ (4), were synthesized by using SPPS chemistry. The synthetic peptides were purified by using RP-HPLC and characterized by using mass spectrometry and CD spectroscopy. Their amyloidogenic nature was determined by using transmission electron microscopy (TEM). We wanted to establish if the designed peptides would inhibit fibrillization of the A_B peptide. We first examined whether the peptides (A β , 2, 3, and 4) would form fibrils spontaneously by using TEM (Figure 4). As expected, $\mathsf{A}\beta$ demonstrated amyloid-like, long, unbranched fibrils whereas 2 dem-

Figure 4. Electron micrographs of peptides A β , 2, 3, 4 alone, and A β incubated with peptides 2, 3, and 4 separately.

onstrated fibrils, but with a reduced density compared to $A\beta$ alone.[8] Peptide 3 did not demonstrate any fibrillar material. In peptide 4, although no fibrils were seen, a few ribbon-like structures were noticed. There was no change in the appearance of the electron micrographs of the peptides even after long periods of incubation (up to one month). The addition of peptides 2, 3, and 4 to A β had a dramatic effect on the morphology of A β fibrillar aggregates. Incubation of A β with 2 reduced the density of fibrils.^[8] The fibrils were completely absent after incubation of AB with 3. Incubation of preformed $A\beta$ fibrils with peptide 3 led to complete disintegration of the fibril structure (data not shown). These results indicate that 3 can be a potent fibrillogenesis inhibitor, and might have a more potent antifibrillizing activity than the native peptide 2. After incubation of A β with peptide 4, the fibrils were less numerous and more thread-like; this suggests that 4 can also partially prevent fibrillization.

CD spectra of the peptides were acquired in trifluoroethanol (TFE; Figure 5). Peptide 2 displayed a relatively strong negative band near 200 nm that was accompanied by a weak negative band or shoulder at \sim 223 nm. This type of CD curve (class U spectrum) is generally characteristic of aperiodic conformers.^[9] The CD spectrum of peptide 3 revealed a positive band at \sim 195 nm and two minima at \sim 228 and \sim 205 nm; the CD spectrum of peptide 4 revealed a positive band at \sim 195 nm and

Figure 5. CD spectra of peptides 2, 3, and 4 in trifluoroethanol.

two minima at \sim 228 and \sim 210 nm. This type of spectrum is suggestive of 3_{10} -helical structure. The CD studies are suggestive of helical structures for peptides 3 and 4. In this regard, Aib-containing helical peptides have been shown to not only disrupt β -sheet structures in a large number of peptides, but also inhibit fibrillization.^[2a-c]

The work by Soto et al. has demonstrated the remarkable potential of pentapeptide, LPFFD, in inhibiting fibrils formed by $A\beta(1-40)$ and $(1-42)$; LPFFD exhibits high solubility, and low immunogenicity.^[1c] However, one major drawback with the use of peptides as drugs in neurological diseases is their rapid metabolism by proteolytic enzymes.^[10a,b] The ability of Δ Phe-containing peptides to resist enzymatic (chymotrypsin) degradation has been shown by using RP-HPLC, in which a single intact peak of similar intensity was seen before and after incubation of peptides 3 and 4 with the enzyme (data not shown). In this context, the stability of Δ Phe residue to enzymatic degradation could be an added advantage.^[11] Moreover, the potential of Δ Phe residue to control not only A β amyloid aggregation but also inhibit amyloid formation by other polypeptides (hIAPP) has been observed (unpublished results). Taken together, these results suggest a new strategy for the development of amyloid-formation inhibitors by using Δ Phe as a β breaker residue. This finding might be of relevance to chemists and biochemists in designing peptide-based antifibrillization drugs that are resistant to enzymatic degradation in vivo.

Experimental Section

Peptide synthesis: Pentapeptide 1 was synthesized by standard solution-phase procedures by using Boc chemistry. Amino acid couplings were performed by using the mixed-anhydride method. Trifluoroacetic acid (TFA) was used to remove the N-terminal Boc group in the peptide fragments. The dehydration of $D,L-\beta$ -phenylserine was carried out by using fused NaOAc and Ac₂O to obtain the Δ Phe moiety. All reactions were monitored by thin layer chromatography (TLC) on precoated silica plates by using a CHCl $_3/$ MeOH (9:1).

Boc-Ile-(b-OH)-Phe-OH: N-methylmorpholine (2.62 mL, 20 mmol) and IBCF (2.78 mL; 20 mmol) were added to a precooled solution $(-20 °C)$ of Boc-Ile-OH (4.62 g, 20 mmol) in THF (30 mL). After being stirred for 30 min, a precooled solution of $D, L-P$ he-(β -OH) (3.982 g, 22 mmol), NaOH (0.88 g, 22 mmol), and H₂O (30 mL) was added, and the mixture was stirred at 0° C for 1 h, and at room temperature, overnight. The organic solvent was removed under reduced pressure and the aqueous phase was acidified with citric acid to pH 3.0 and extracted with EtOAc $(3 \times 15 \text{ mL})$. The EtOAc layer was washed with H₂O, dried over anhydrous Na₂SO₄ and evaporated to yield a cream-colored, powdery compound; yield: 6.9 g (88%), R_f = 0.3, m.p. 85 °C.

Boc-Ile- Δ Phe-azlactone: Boc-Ile-(β -OH)-Phe-OH (5.91 g; 15 mmol) was treated with anhydrous NaOAc (1.599 g, 19.5 mmol) and freshly distilled Ac_2O (30 mL) for 36 h at room temperature. The mixture was poured over crushed ice; the resultant precipitate was washed with cold H₂O and finally recrystallized from acetone/H₂O; yield: 6.3 g (88%), R_f = 0.6, m.p. 95 °C.

Fmoc-Val/Leu- Δ Phe-azlactone: This was synthesized by using a similar method as described above.

Fmoc-Val-(β **-OH)-Phe-OH:** Yield: 9.17 g (20 mmol; 91.4%), $R_f =$ 0.30, m.p. 72-74 $\,^{\circ}$ C.

Fmoc-Val- Δ **Phe-azlactone**: Yield: 4.33 g (10 mmol; 93%), $R_f = 0.95$, m.p. 102-104 °C.

Fmoc-Leu-(β **-OH)-Phe-OH:** Yield: 9.8 g (20 mmol; 95%), $R_f = 0.36$, m.p. 82-84 °C.

Fmoc-Leu- Δ **Phe-azlactone**: Yield: 4.56 g (10 mmol; 95%), $R_f =$ 0.82, m.p. $164-166$ °C.

Synthesis of peptide 1: The tripeptide Boc-Ile- Δ Phe-Ala-OMe was first synthesized. Ala-OMe·HCl (2.196 g, 15.8 mmol) was added to a solution of Boc-Ile- Δ Phe-Azl (5.656 g, 15.8 mmol) in CH₂Cl₂ (20 mL), then $Et₃N$ (2.15 mL, 15.8 mmol) was added, and the mixture was stirred at room temperature for 30 h. The solvent was removed under reduced pressure, and the aqueous phase was washed once with EtOAc and acidified with solid citric acid. The EtOAc layer was washed with H₂O, dried over anhydrous Na₂SO₄, and evaporated to yield the desired product. The tripeptide Boc-Ile-APhe-Ala-OMe was deprotected at its N terminus by using a mixture of TFA in CH₂Cl₂ (1:1, v/v) at room temperature for 30 min. Excess acid was removed in vacuo, and the residue was triturated with dry $Et₂O$ and filtered. Dicyclohexylcarbodiimide (DCC) (0.42 g, 2.1 mmol) and 1-hydroxybenzotriazole (HOBT) (0.33 g, 2.1 mmol) were added to a precooled solution of Boc-Ala-OH (0.397 g, 2.1 mmol) in DMF (20 mL), and the mixture stirred for 30 min. The trifluoroacetate salt of the tripeptide in DMF and $Et₃N$ (0.28 mL) was added, and the mixture was stirred at room temperature for 16 h. The precipitated dicyclohexylurea was filtered off, and the solvent was removed in vacuo. The reaction was worked up as above to yield the tetrapeptide Boc-Ala-Ile- Δ Phe-Ala-OMe. The Boc group of the tetrapeptide was deprotected by using TFA in CH_2Cl_2 , and the deprotected peptide was coupled with Boc-Phe-OH by using DCC and HOBT as above, to finally yield the cream-colored Boc-Phe-Ala-Ile-DPhe-Ala-OMe. The pentapeptide was purified by RP-HPLC with a Waters C18 column (300 mm \times 3.9 mm) by using a MeOH/H₂O gradient; yield: 700 mg (80%), $R_f = 0.81$ (CHCl₃/MeOH, 9:1), m.p. 165 °C (from MeOH). Peptide identity was confirmed with ES-MS; m/z calcd for $C_{36}H_{49}N_5O_8$: 679; found: 702 [M+Na]⁺.

Synthesis of peptides 2, 3, and 4: Fmoc-protected amino acids (Fmoc: 9-flourenylmethyoxycarbonyl) for solid-phase peptide syn-

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thesis were obtained from Novabiochem (India). The three pentapeptides were synthesized manually on a 0.5 mmol scale. Fmocrinkamide p-methylbenzhydrylamine (MBHA) resin (0.5 mmol g^{-1}) was used to afford the C-terminal primary amides. Couplings were performed by using carbodiimide. Solution-phase methodology was used to introduce the Δ Phe residue in the pentapeptides as a dipeptide block by dehydration of Fmoc-aa-D,L-threo-β-phenylserine (where aa is either valine or leucine) by using fused NaOAc and freshly distilled Ac₂O as reported earlier. Fmoc deprotection was performed with piperidine (20% in DMF). After the addition of the final residue, the N terminus was acetyl capped, and the resin was rinsed with $DMF/CH_2Cl_2/MeOH$, and then dried. The final peptide deprotection and cleavage from the resin was achieved with TFA/ H2O/TIS 95:2.5:2.5 (10 mL) for 2 h. The resin was filtered and the filtrate was concentrated, after which the crude peptides were precipitated with cold diethyl ether. The supernatant was decanted, and the crude peptides were suspended in H_2O with a minimal amount of acetonitrile (ACN), then frozen and lyophilized to dryness. The crude peptides were purified by reverse-phase HPLC by using a TFA (0.1%) in H₂O/ACN on a Waters Deltapak C18 (19 mm \times 300 mm). A linear gradient of ACN from 5 to 80% over 80 min at a flow rate of 6 mLmin $^{-1}$ was used. Physical characterization of the synthesized peptides was as follows: 2: yield: 325 mg (94%), m.p. 170-175°C, t_R 54 min, m/z 693.5 Da; 3: yield: 304 mg (88%), m.p. 180-185 °C, t_{R} 57 min, m/z 691.2 Da; 4: yield: 307 mg (89%), m.p. 189–194 °C, $t_{\rm R}$ 57 min, m/z 691.3 Da.

X-ray crystallography: The purified peptide was crystallized from MeOH by using the slow evaporation method over a period of three months. The X-ray diffraction data were collected by using a Bruker APEX CCD diffractometer equipped with Mo-K_a radiation. The structure was solved by using the direct method employed in the SHELXS 97 program. The refinement of the structure was carried out by using full-matrix least-squares refinement on F_o^2 implemented in SHELXL 97 software.

Transmission electron microscopy: The synthetic AB_{1-40} was dissolved in H₂O prior to dissolution in PBS buffer (A β has low solubility in buffers) to produce a 0.5 mm stock solution. Lyophilized peptide ligands were dissolved in 100% hexafluoroisopropanol (HFIP) at a concentration of 100 mm. To avoid preaggregation, fresh stock solutions were prepared for each experiment. Peptide stock solutions were diluted with PBS to a final concentration of 10 mm. The $\mathsf{A}\beta$ peptide sample was incubated either with or without peptide ligands for 10 days at room temperature. A sample (10 mL) was placed on a 400-mesh copper grid covered with carbon-stabilized formvar film. After 1 min, the excess fluid was removed and the grid was negatively stained by treatment with 2% uranyl acetate in $H₂O$ for 2 min.

Circular dichroism studies: The spectra were acquired between 190–330 nm (0.1 cm cell, peptide concentration \sim 100 mm) at 0.1 nm intervals with a time constant of 4 s and a scan speed of 200 nm min $^{-1}$, and averaged over six separate scans. The spectra were baseline corrected and smoothed. The peptide concentration was determined by using the molar extinction coefficient of Δ Phe $(-19000 \text{ m}^{-1} \text{cm}^{-1}).$

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